

Unusual Edge Shift of Stepwise-Reduced Cobalt Tetraphenylporphyrin and Tetrabutylphthalocyanine

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Introduction: We have shown that metallo-porphyrins and phthalocyanines ($M = \text{Co}$ and Fe) are effective photocatalysts in the reduction of CO_2 to CO and formate [1-3]. Metal(I) tetraphenylporphyrins ($\text{M}^{\text{I}}\text{TPP}$) and metal(I) tetrabutylphthalocyanines ($\text{M}^{\text{I}}\text{TBPC}$) do not react with CO_2 until they are reduced beyond the $\text{M}(\text{I})$ state. The site of reduction, metal *versus* macrocycle, along with changes in the catalyst's structural and electronic properties are important in understanding the active species in these catalytic reactions. Therefore we have recently measured the XANES of a series of CoTPP and CoTBPC .

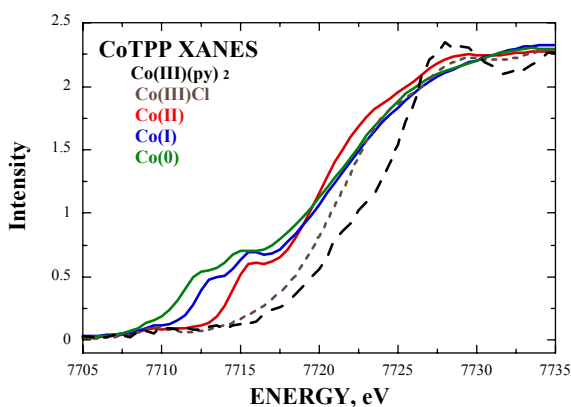
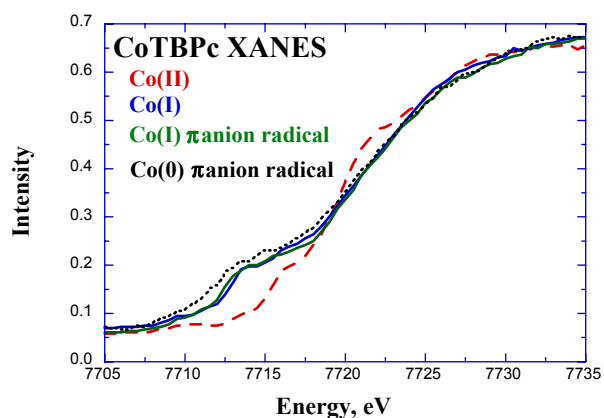
Methods and Materials: Stepwise reduction of $\text{Co}^{\text{II}}\text{TPP}$ and $\text{Co}^{\text{II}}\text{TBPC}$ were carried out in CH_3CN using Na-Hg in homemade cells. The purity of these air-sensitive samples was examined by measuring optical spectra before and after the XANES measurements. The decomposition of samples is typically much less than 10 %. XANES of $[\text{Co}^{\text{III}}\text{TPP}(\text{Cl})]$ and $[\text{Co}^{\text{III}}\text{TPP}(\text{py})_2]^+$ ($\text{py} = \text{pyridine}$) were also measured.

Results: XANES results of CoTPP and CoTBPC are shown in Figures 1 and 2, respectively. In $\text{Co}^{\text{III}}\text{TPP}^+$, the chloride ion donates more electron density to the cobalt(III) center than pyridine, and partially reduces the metal, resulting in a -1 eV edge shift. It is striking that XANES spectra for $\text{Co}^{\text{II}}\text{TPP}$, $[\text{Co}^{\text{I}}\text{TPP}]^-$, and $[\text{Co}^0\text{TPP}]^{2-}$ exhibit no main-edge shift. However the pre-edge ($1s \rightarrow 4p_z$) positions for $\text{Co}^{\text{II}}\text{TPP}$, $[\text{Co}^{\text{I}}\text{TPP}]^-$, and $[\text{Co}^0\text{TPP}]^{2-}$ are sensitive to stepwise reduction. The main-edges for $\text{Co}^{\text{II}}\text{TBPC}$, $[\text{Co}^{\text{I}}\text{TBPC}]^-$, $[\text{Co}^{\text{I}}\text{TBPC}^*]^{2-}$ and $[\text{Co}^0\text{TBPC}^*]^{3-}$ also indicate no shift, however, the pre-edges are sensitive to the oxidation state of the cobalt centers (see Figure 2). The lack of edge shifts upon reduction of $\text{Co}^{\text{II}}\text{TPP}$ and $\text{Co}^{\text{II}}\text{TBPC}$ suggests that considerable electron density resides in the ligand π system. However, the pre-edge shifts are consistent with the assignments from the optical spectra. In the case of CoTPP , the first and second reductions of $\text{Co}^{\text{II}}\text{TPP}$ are metal-centered to form $[\text{Co}^{\text{I}}\text{TPP}]^-$ and $[\text{Co}^0\text{TPP}]^{2-}$, respectively. In the case of cobalt phthalocyanines, the first, second and third reductions of $\text{Co}^{\text{II}}\text{TBPC}$ are metal-, ligand- and metal-centered reductions, respectively, forming $[\text{Co}^{\text{I}}\text{TBPC}]^-$, $[\text{Co}^{\text{I}}\text{TBPC}^*]^{2-}$ and $[\text{Co}^0\text{TBPC}^*]^{3-}$. Origin of pre-edge shift and doubling of the $1s \rightarrow 4p_z$ edge (in $\text{Co}(\text{I})$ and $\text{Co}(\text{0})$ species) are under investigation.

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Figure 1. XANES of CoTPP Figure 2. XANES of CoTBPC